## **Playing With Polymers**

Name \_\_\_\_\_

1. What is polymerization? It is the process of chemically bonding \_\_\_\_\_\_ to form \_\_\_\_\_\_.

### 2. What are some examples of natural polymers?

- Cotton, silk, wool, and natural rubber are all \_\_\_\_\_\_ polymers made from plant or animal products.
- The \_\_\_\_\_\_found in our cells is also a polymer. The monomer units of DNA are \_\_\_\_\_\_ and the polymer is known as a polynucleotide.

### 3. What are some examples of synthetic (manmade) polymers?

- Some synthetic polymers are manmade \_\_\_\_\_\_ and fabrics such as polyester, \_\_\_\_\_, and rayon.
  is a synthetic polymer used to make a variety of products, such as water bottles, plastic
- \_\_\_\_\_\_, and kitchen utensils as well as safety \_\_\_\_\_, computers, and rocket engines.
- Polymers are also used in \_\_\_\_\_\_ as substitutes for human \_\_\_\_\_\_, such as bones and arteries.

#### 4. What substances are used to make synthetic polymers?

They are made from materials called _	,	which are made from	Petroleum is
made from the remains of	_ and	_ that have been buried benea	th sediments at the bottom of
the ocean. Over time,	and	changed the remains	into petroleum, which is a
resource .			

#### 5. Answer these questions about measuring volume.



#### 6. Answer these questions about measuring mass.

What unit is used to represent mass?

What is the mass of the object measured in the picture?



### 7. Fill in each blank as you discuss the safety rules in class.

(1) Read \_\_\_\_\_\_ carefully! If you are not sure what to do, ask for help. Do not make up your own recipes!

(2) \_\_\_\_\_\_ should always be worn when experimenting with chemicals.

(3) Do not \_\_\_\_\_\_ directly from the container! If you need to sniff slime, hold it several inches away from your face and use your hand to wave fumes towards your nose.

(4) No eating or drinking during the lab, which also means that you should NOT eat the slime or \_\_\_\_\_\_ any substances used to make slime! Keep your slime out of reach of small \_\_\_\_\_\_ and \_\_\_\_\_.

(5) Do not put the slime where it doesn't belong, such as on \_\_\_\_\_, carpeting, or other people!

(6) Dispose of slime materials properly. All slime must be thrown away in the trash can. Use a dry towel to clean your hands, cup, and plate. <u>DO NOT put any amount of slime in the \_\_\_\_\_</u>!

(7) Clean up \_\_\_\_\_\_ immediately! Your lab area should be clean when you start and clean when you leave.

(8) \_\_\_\_\_ your hands before you leave class.

(9) \_\_\_\_\_! No hitting, shoving, or other horseplay is allowed!

(10) Slime must remain in the \_\_\_\_\_! You are not allowed to take it to other classes. You will be able to take the slime home on the last day!

(11) Most of the slime will keep for \_\_\_\_\_ days. After your slime goes bad, throw it away! Do not dump in a sink!

(12) If you do not follow the rules, you will not be allowed to do the experiments and will earn a \_\_\_\_\_\_ grade for this unit. If you agree to follow these safety rules, sign your name in the shaded box below.

I agree to follow the safety rules. I understand that if I do not follow the rules, I will not be allowed to do any experiments and will receive a zero grade for this unit.

Name \_\_\_\_\_

\_ Date \_

# **Dlaying With Dolymers**

Name \_



### Across

2. Thickening agent used in many foods (or in goobers)

4. Synthetic polymer used to make many products, such as water bottles and toys

5. A natural polymer created by many caterpillars and spiders

7. Links in a polymer chain

9. Manmade polymers, such as plastic, nylon, and polyester

- 12. Instrument used to measure mass
- 16. Used to help prevent wrinkles in your clothes

17. Chain of monomers that are chemically bonded together

18. A synthetic polymer used to cover food to prevent spoilage and contamination

19. The code of life found in each of our cells

22. Polymers made from plant and animal products found in nature, such as cotton, wool, silk, and natural rubber

23. Monomer that makes up the DNA molecule

24. Used as a laundry agent; sodium tetraborate

### Down

- 1. Process by which polymers are formed
- 2. Instrument used to measure volume
- 3. Slime made from laundry starch and white glue
- 4. Substance in white glue; used to make Super Slime
- 5. Slime made from PVA and borax solution
- 6. A slime made from guar gum and borax solution
- 8. Building blocks of proteins
- 10. A natural polymer used to make "soft" clothing
- 11. A mixture made from water and corn starch
- 13. A common adhesive used for making slime

14. Polymers may be used in this field as substitutes for human tissues

15. A synthetic polymer that is used to make silky fabrics, such as pantyhose

18. Polymer made up of chains of amino acids; also called a polypeptide chain

20. A natural polymer used for clothing, such as coats and socks

21. Slime made from white glue and borax solution

## How could you describe your slime-making experience?

Find 14 words in the puzzle below that could be used to describe your experience! Write the words on the lines at the bottom of this page.



# **Playing With Polymers**

# **Data Chart**

Test	Gloop	Boogers	Goobers	Super Slime
<b>Description</b> Color, texture, odor, or other observations				
Slime Rating 1 = not very slimy to 4 = very slimy				
<b>Slow Poke Test</b> Slowly poke your finger into the slime. What happens?				
<b>Quick Poke Test</b> Quickly poke your finger into the slime. What happens?				
<b>Slow Pull Test</b> Slowly pull on the ends of a piece of the slime. What happens?				
<b>Quick Pull Test</b> Quickly pull on the ends of a piece of the slime. What happens?				
<b>Blob Test</b> Roll your slime into a ball and let it sit for a minute. What happens?				
Hang Test How long does it take for the slime to reach the table from a height of 30 cm?				
<b>Bounce Test</b> Roll into a ball and drop it on the table. Rate the bounce – 1 – poor to 5 -great!				

## **Slime Tests**

### Description

What does the slime look like? Does it smell? How would you describe its texture/feel?

## **Slime Rating**

How slimy is your slime? Rate it from 1 = not very slimy to 4 = very slimy.

## **Slow Poke Test**

Roll the slime into a ball, and then <u>slowly</u> poke your finger into it. What happens? How far does your finger go into the slime?

### **Quick Poke Test**

Roll the slime into a ball, and then <u>quickly</u> poke the slime with your finger. What happens? How far does your finger go into the slime?

NOTE: If you are not able to do a test (slime too runny or plops), write a note in that space on your chart to explain why you were not able to do the test.

## **Slow Pull Test**

Grab a glob of goop with your fingers and slowly pull on the ends. What happens?

## **Quick Pull Test**

Grab a glob of goop with your fingers and quickly pull on the ends. What happens?

### **Blob Test – Need a timer!**

Roll your goop into a ball and then sit the ball of slime on your plate or the table and time how long it takes for it to be a "blob" or flatten out. Write the time in your data chart.

### Hang Test - Need a timer and a ruler!

Hold a glob of slime at a height of 30 cm above the table. Time how long it takes for the slime to reach the table. Write the time in your data chart.

### **Bounce Test - Need a ruler!**

Roll your goop into a ball and drop from a height of <u>30 cm</u> above the table. What happens?

T. Tomm 2008 Updated 2024 <u>https://sciencespot.net/</u>